## PLANT ANTITUMOUR AGENTS. XII. TEXENSINE, A NEW PEPTIDE ALKALOID ${\tt FROM\ COLUBRINA\ TEXENSIS}^{1,2}$

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Recently, a number of peptide alkaloids have been isolated from the plants of the Rhammaceae family.<sup>3,4</sup> In the course of our continuing study of Colubrina texensis (Gray) (Rhamnaceae) for tumour inhibitory substances, we have isolated a new peptide alkaloid texensine (1).<sup>5</sup>

The alcoholic extract of the dried aerial parts of the plant was partitioned between chloroform and water and the residue from the chloroform layer was partitioned between 10% aqueous methanol and petroleum ether. Chromatography of the residue from the aqueous methanol layer on Florisil using 25% methanol in chloroform followed by repeated preparative tlc on silica gel (15% acetone in chloroform) and crystallization from ethyl acetate gave texensine (1) (0.0005%),  $M^{+}$  at m/e 573.3318, calcd. for  $C_{33}H_{43}N_{5}O_{4}$ , m/e 573.3315; mp 249-252°;  $[\alpha]_{D}^{25}$  - 144° (C 0.50, CHCl<sub>3</sub>). The infrared spectrum of texensine showed absorption bands at 3475, 3385 (NH), 1685 (amide I) and 1495 (amide II) cm<sup>-1</sup> suggesting a peptide.

The structure of texensine was deduced from its mass spectrum (Figure 1) which showed the characteristic peptide decomposition pattern. The molecular compositions of fragment ions were determined by high-resolution mass spectroscopy. Thus the most intense peak at m/e 114 ( $^{C}_{7}H_{16}N$ , Chart 1) corresponding to the ion 4 indicated that the basic terminal amino acid was N,N-dimethyl-leucine (5). This was further supported by the ion at m/e 516 (M-C<sub>4</sub>H<sub>9</sub>) and the presence of nmr signals at  $^{6}_{2}$  2.27 [S,  $^{6}_{2}$  N(CH<sub>3</sub>)<sub>2</sub>] and 0.99 [d,  $^{6}_{2}$ ,  $^{6}_{3}$  HC(CH<sub>3</sub>)<sub>2</sub>].

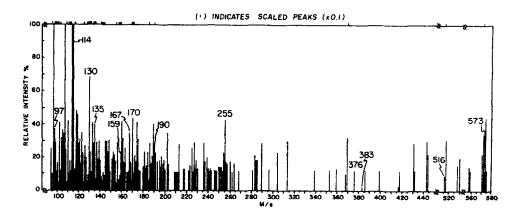


Figure 1. Mass Spectrum of Texensine

The peaks at m/e 130 ( $C_9H_8N$ ), 159 ( $C_{10}H_{11}N_2$ ), and 170 ( $C_{11}H_8N0$ ) corresponded to ions  $\underline{6}$ ,  $\underline{7}$ , and  $\underline{8}$ , respectively and are typical for <u>tryptophan</u> (9) containing peptide alkaloids. The presence of tryptophan molety was further confirmed by the absorption maxima at 281 and 290 nm in the ultraviolet spectrum of texensine ( $\underline{1}$ ).

The peak at m/e 97 ( $C_6H_90$ ) corresponding to the ion  $\underline{10}$  suggested the presence of  $\beta$ -hydroxy-leucine ( $\underline{11}$ ) moiety in the molecule. This was further substantiated by the presence of a pair of doublets ( $J = 6H_2$ ) centered at  $\delta 0.89$  and 1.24 due to the methyl groups of this moiety.

Finally the peak at m/e 135 (C<sub>8</sub>H<sub>9</sub>NO) corresponding to the ion <u>12</u> suggested the presence of <u>p-hydroxystyrylamine</u> (18) moiety which is characteristic of these peptide alkaloids.<sup>4</sup>

The joining of the four units--N,N-dimethylleucine (5), tryptophan (9),  $\beta$ -hydroxyleucine (11), and p-hydroxystyrylamine (18)--was evident from other fragmentation peaks. Thus the peak at m/e 190 ( $C_{12}H_{16}NO$ ) corresponding to the ion 13 showed that the phenolic oxygen was etherified with  $\beta$ -hydroxyleucine (11). The ion at m/e 376 ( $C_{23}H_{26}N_3O_2$ ) corresponding to the formula 14 showed that the styrylamine was acylated by the tryptophan carboxyl. The two ions at m/e 283 ( $C_{17}H_{19}N_2O_2$ ) and 255 ( $C_{16}H_{19}N_2O$ ) represented by formulas 15 and 16 respectively indicated that  $\beta$ -hydroxyleucine and tryptophan were directly linked to complete the macrocyclic ring. The ion at m/e 167 ( $C_{9}H_{15}N_2O$ ) corresponding to the formula 17 proved that N,N-dimethylleucine was attached to the macrocyclic ring through the nitrogen of  $\beta$ -hydroxyleucine.

In agreement with the assigned structure hydrogenation of  $\underline{1}$  gave dihydrotexensine ( $\underline{2}$ ) (Required for  $C_{33}H_{45}N_5O_4$ : m/e 575. Found: m/e 575) which upon acid hydrolysis gave p-tyramine.

## Chart 1

It appears that texensine is an N-methylated derivative of homoamericine (3). It should, however, be pointed out that the latter has only been reported as a contaminant (4%) of americine and has never been isolated in pure state.

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## REFERENCES

- For previous papers in this series see: M. C. Wani, H. L. Taylor and M. E. Wall,
  J. C. S. Chem. Comm., 390 (1973).
- This investigation was conducted under contract NIH-69-2019 from Division of Cancer Treatment, National Cancer Institute, National Institutes of Health, DHEW.
- 3. R. Tschesche, E. U. Kaussmann and H.-W. Fehlhaber, Tetrahedron Letters, 9, 865 (1972).
- 4. E. W. Warnhoff, Fortschr. Chem. Org. Naturstoffe, 28, 162 (1970).
- 5. Texensine did not exhibit any cytotoxicity (ED<sub>50</sub>) against KB cell culture at 10 μg/ml, nor P388 activity at 8 mg per kilogram level. Cytotoxicity and in vivo activity were carried out under the auspices of the National Cancer Institute by the procedures described in Cancer Chemotherapy Reports, 25, 1 (1962).
- 6. H.-W. Fehlhaber, Z. Analyt. Chem., 235, 91 (1968).
- 7. P. Pfander, Liebigs Ann. Chem., 707, 209 (1967).
- 8. R. E. Servis, A. I. Kosak, R. Tschesche, E. Frohberg and H.-W. Fehlhaber, J. Amer. Chem. Soc., 91, 5619 (1969).
- 9. F. K. Klein and Henry Rapoport, J. Amer. Chem. Soc., 90, 2398 (1968).